

Sample preparation and analysis of Brominated Flame Retardants (BFR) in environmental samples

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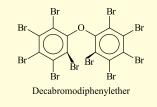
Summary

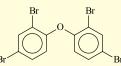
- Method for cleanup and analysis of BFRs in environmental samples
- Detection and quantification of a wide range of concentration of BFRs
- Several BFRs with very different chemical and physical properties are separated and quantified

Introduction

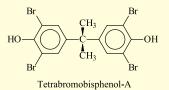
Brominated flame retardants are applied to several materials for their flame retardant properties

- Electrical and electronic equipment
- Textiles, coatings; sofas, seats of cars,
- buses, and aircraft
- Plastics
- Building materials
- Paints





Tetrabromodiphenylether





Hexabromocyclododecane



Disodiumtetrabromphtalate

The BFRs have very different structures and characteristics. The most commonly used BFRs are the brominated diphenylethers (PBDE), hexabromocyclododecane (HBCD) and tetrabromobisphenol-A (TBBPA). Their presence in the environment has become a matter of great concern

- Highly persistent
- Accumulate in the environment
- Atmospheric long-range transport
- Found in animals on top of the trophic level
- Suspected to be harmful during longterm exposure
- Effects resemble the toxic effects of PCBs.

Method

Sample types

- Biological samples (biological oils, fish, mussel, moss, etc.)
- Soil and sediment
- Sewage

• Water (waste water, drinking water etc.)

Cleanup and preparation

- Addition of internal standards
- Solvent extraction
- Gel Permeation Chromatography
- Fractionated silica chromatography
- Derivatisation; allowing hydroxylated compounds to be separated on GC
- Addition of recovery standard to ensure the quality



The instrument is a Gas Chromatograph coupled with a High Resolution Mass Spectrometer (GC/HRMS) used for analysis of PBDEs, TBBPA, BPA.

For analysis of HBCD a High resolution Liquid Chromatography coupled with Mass Spectrometry (HRLC/HRMS) is used. This allows separation of all three isomers (α , β and γ -HBCD).



BFRs are also used in electronic parts of colour televisions and personal computers.

	Chemical name
BDE-28	2,4,4'-tribromodiphenylether
BDE-47	2,2',4,4'-tetrabromodiphenylether
BDE-71	2,3',4',6- tetrabromodiphenylether
BDE-99	2,2',4,4',5-pentabromodiphenylether
BDE-100	2,2',4,4',6-pentabromodiphenylether
BDE-119	2,3',4,4',6-pentabromodiphenylether
BDE-138	2,2',3,4,4',5'-heksabromodiphenylether
BDE-153	2,2',4,4',5,5'-heksabromodiphenylether
BDE-154	2,2',4,4',5,6'-heksabromodiphenylether
BDE-183	2,2',3,4,4',5',6-heptabromodiphenylether
BDE-209	Decabromodiphenylether
TBBPA	Tetrabrombisphenol-A
BPA	Bisphenol A
α-HBCD	α-hexabromocyclododecan
β-HBCD	β- hexabromocyclododecan
γ-HBCD	γ- hexabromocyclododecan

Brominated compounds separated, identified and quantified by the method developed.

Conclusions

• Separation and identification of several BFRs in different types of samples

- Suitable for all levels of contamination in several different environmental samples
- Proven useful for locating sources of contaminations

Findings has shown various accumulative properties for different isomers.

References

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